

①②

EUROPEAN PATENT APPLICATION

②① Application number: 83302539.8

⑤① Int. Cl.³: **B 01 J 8/06, C 07 D 301/08**

②② Date of filing: 05.05.83

③① Priority: 07.05.82 US 375795

⑦① Applicant: **THE BABCOCK & WILCOX COMPANY**,
1010 Common Street P.O. Box 60035, New Orleans
Louisiana 70160 (US)

④③ Date of publication of application: 16.11.83
Bulletin 83/46

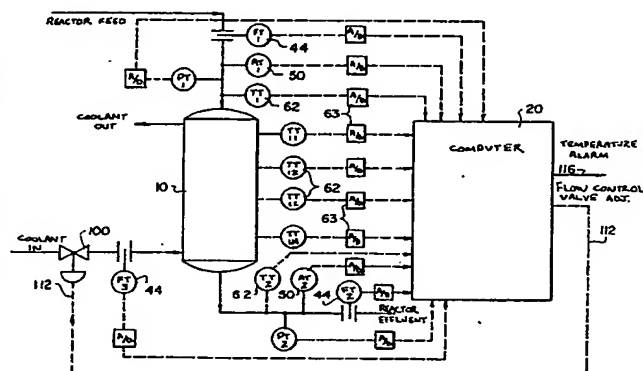
⑦② Inventor: **Agarwal, Suresh C.**, 26011 Lakeshore
Boulevard, Euclid Ohio 44132 (US)

⑧④ Designated Contracting States: **DE FR GB IT**

⑦④ Representative: **Cotter, Ivan John et al, D. YOUNG & CO.** 10 Staple Inn, London WC1V 7RD (GB)

⑤④ **Reactor temperature control systems.**

⑤⑦ A temperature control system controls the temperature of an olefin oxidation reactor (10) by measuring various parameters in the reactor and in flow lines to and from the reactor, and using these parameters with known quantities for specific heat of the feed and effluent and heats of vaporization and reaction, to calculate a coolant flow rate set point. Parameters relating to the heat balance with regard to a desired olefin oxide product and undesired carbon dioxide products are utilized for correct control. Equipment is also provided for ascertaining the state of a catalyst used in the reaction. Circuit components are provided for obtaining changes in flow rates, temperatures and concentrations to calculate a change in coolant flow rate as well as the steady state effluent flow rate quantity to improve the system response to start up, shut down and transient conditions.



REACTOR TEMPERATURE CONTROL SYSTEMS

This invention relates to reactor temperature control systems or arrangements, and is related to our earlier European Patent Application No. 83301958.1 (Publication No. EP _____), hereinafter referred to as "the earlier application". Systems embodying the invention may, for example, be used for controlling the temperature of an olefin oxidation reactor by regulating the rate of coolant flow to maintain the reactor temperature within a desired temperature range, during the operation of the reactor as well as during start-up, shut-down and transient operating conditions.

Various techniques and systems are known for controlling chemical reactors.

In addition to the non-anticipating but relevant patents set forth in the earlier application, U.S. Patent No. 3 471 582 to Lupfer discloses an arrangement wherein a desired temperature across an exothermic reactor is maintained by controlling the reactor feed temperature in response to a difference in temperature between the reactant feed and product stream from the reactor until a maximum predetermined product temperature is obtained.

U.S. Patent No. 3 271 472 to Ogle et al discloses apparatus for controlling the operation of a thermal cracking furnace. Since the thermal cracking of hydrocarbons is an endothermic reaction, it is necessary to maintain a maximum possible temperature within the equipment limits. A minimum temperature is not considered or important in Ogle et al.

See also U.S. Patent No. 4 249 907 to Callegas which discloses a selective hydrogenation process wherein at least one catalyst is utilized. The temperature of feed stream to a catalyst bed is controlled so as to maintain a desired reaction temperature in the catalyst bed.

In an olefin, in particular ethylene, oxide manufacturing process, ethylene and oxygen or air are mixed and fed to an isothermal multitubular reactor. Ethylene is oxidized into ethylene oxide in the presence of a catalyst and carbon dioxide and water are produced as by-products. Reactor temperature control objectives are:

operation at the most economical temperature;
operation within a safe zone;
maximum conversion to ethylene oxide while minimizing by-products;
reduction of consumption of coolant;
5 avoidance or elimination of unsafe operation; and
reduced operator attention.

Reactor temperature control is of key significance because of the following factors:

1. The most economical temperature for oxidation is one at which
10 occurs the highest conversion to ethylene oxide rather than to by-products.
2. Catalyst selectivity increases as the reaction temperature is lowered while ethylene conversion increases with increasing reactor temperature. Thus, temperature requirements for high selectivity and
15 high conversion are opposed. This results in a narrow temperature range for reactor operation.
3. Increase in reaction temperature produces two effects: (1) the overall rate of ethylene oxidation increases, and (2) catalyst selectivity to ethylene oxide decreases such that relatively more ethylene is
20 converted into carbon dioxide and water. Moreover, heat generation increases due to the fact that more ethylene is oxidized and the overall reaction becomes less selective. Consequently, increase in temperature may result in:
 - a reactor runaway condition;
 - 25 catalyst poisoning;
 - increased coolant demand;
 - an unsafe operating situation; and/or
 - increased operator attention.

Thus, neither a temperature rise nor a temperature drop is desirable.

- 30 In a state of the art system, reactor temperature control is based on manipulating coolant flow rate. A set point of the system is directly based upon average reactor temperature. Such a control scheme results in almost all of the deficiencies described above.

- According to the present invention there is provided a system for
35 controlling the temperature of a reactor for containing a reaction from at least one reactant to at least another product, the reactor having a feed line

for the reactant and an effluent line for the product, the arrangement being characterised by:

a feed flow transmitter connected to the feed line for measuring the flow of reactant to the reactor;

5 an effluent flow transmitter connected to the effluent line for measuring the flow of product from the reactor;

a feed temperature transmitter connected to the feed line for sensing the reactant temperature;

10 an effluent temperature transmitter connected to the effluent line for measuring the product temperature;

at least one reactor temperature transmitter connected to the reactor for measuring a temperature of the reactor;

a concentration transmitter connected to the effluent line for measuring the concentration of the at least one product in the effluent
15 line;

a coolant flow line to the reactor for supplying coolant to the reactor at a coolant flow rate;

coolant flow control means in the coolant line; and

circuit means that

20 (i) is connected to all of said transmitters and to the coolant flow control means for controlling the flow of coolant to the reactor according to a coolant flow signal,

(ii) is connected to receive quantities proportional to the heat of reaction for at least one reaction in the reactor, specific heats of the reactant and
25 product, and the heat of vaporization of the coolant,

(iii) is operable to obtain values for changes per unit time in feed flow rate, effluent flow rate, feed temperature, reactor temperature, effluent temperature, and concentration of at least one product, and

30 (iv) includes circuit components for multiplying each change per unit time by a characteristic factor.

A preferred embodiment of the present invention described in detail hereinbelow comprises a control arrangement or system which permits operation of an olefin oxidation reactor at the most economical and safe temperature range, with regard to a maximum conversion of the olefin to
35 the desired olefin oxide and a minimization of by-products. The preferred control system is in particular applicable to ethylene oxidation reactors but is also applicable to other exothermic and endothermic reactors.

According to the preferred embodiment of the invention, a system is provided which controls the rate of coolant flow in the chemical reactor according to an algorithm which incorporates various parameters including reactor feed and effluent flow rates, specific heat of reactants and products, reactor and effluent temperatures, coolant heat of evaporation, reactant and product concentration and heat of reactions for various reactions taking place in the reactor. In addition, temperatures are taken at varied locations along the reactor length, for obtaining a maximum and a minimum value for temperatures within the reactor for establishing a desired reactor temperature range. The preferred temperature control system is simple in design, rugged in construction and economical to manufacture.

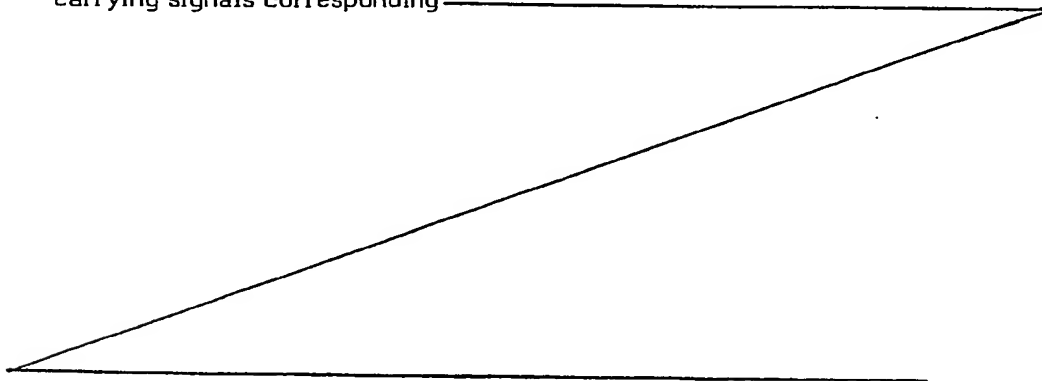
The preferred control system or arrangement, which operates by controlling the rate of coolant flow, additionally compensates for variations in operating conditions as well as providing proper coolant flow control even during start-up and shut-down phases of the operation.

The invention will now be further described, by way of illustrative and non-limiting example, with reference to the accompanying drawings, in which:

Figure 1 is a schematic representation of a temperature control system embodying the invention in combination with a tubular reactor for containing an olefin oxidation reaction; and

Figure 2 is a block diagram which illustrates an exemplary computer layout for implementing the system.

Figure 1 shows a control system or arrangement for controlling the flow of coolant into a tube reactor 10 by controlling the position of a valve 100 in a coolant inlet line. Control of the valve 100 is achieved over a line 112 connected to the output of a computer 20 which receives various inputs carrying signals corresponding



to temperatures at various locations in the reactor and in the reactor feed at effluent lines and component concentrations. The computer is programmed with constants which relate to heats of various reactions going on in the reactor 10 and various physical characteristics of the coolant, the reactants and the products of the reaction.

The control system described in the earlier application is based on the following assumptions:

- a. Specific heat of reactor feed and effluent streams is assumed to be constant with changes in temperature and steam composition.
- b. Heat of reaction is independent of temperature.

The above assumptions, however, while usually valid, do not hold during start-up, shut-down and transient reactor operating conditions. Consequently, coolant flow rate may be less than the desired rate, whereby the reactor will operate at a temperature higher than necessary. The present control system takes these variations into account.

As shown in the earlier application, coolant flow rate is given by the expression

$$Q = \frac{1}{\lambda} [F_2 \{y_1 \Delta H_1 + y_2 \Delta H_2 + C_{P1} (T_R - T_0)\} - F_1 C_{P1} (T_R - T_I)] \quad (1)$$

where:

Q = Flow rate of coolant;

λ = Heat of vaporization for coolant;

F_1 = Flow rate of feed;

$$C_{P1} = \sum_{k=1}^4 C_{P_k} x_k \quad (\text{for feed})$$

C_{P_k} = specific heat of component k in feed; ($k = 1$ for ethylene, 2 for carbon dioxide, 3 for ethylene oxide and 4 for oxygen);

X_k = Concentration of component k in feed;

5 T_R = Reaction temperature;

T_I = Feed inlet temperature;

T_O = Reactor Exit Stream temperature;

$$C_{P_1} = \sum_{m=1}^4 C_{P_m} y_m \quad (\text{for effluent})$$

10 C_{P_m} = specific heat of component m in effluent; ($m = 1$ for ethylene oxide, $m = 2$ for CO_2 , $m = 3$ for ethylene, $m = 4$ for water);

ΔH_1 = Heat of reaction for oxidation to ethylene oxide;

ΔH_2 = Heat of reaction for oxidation to carbon dioxide and water;

15 F_2 = Flow rate of reactor effluent;

y_1 = a first produce (ethylene oxide) concentration; and

y_2 = a second produce (CO_2) concentration.

Functionally, equation (1) can be written as

$$Q = f[F_1, F_2, T_I, T_R, T_O, y_1, y_2] \quad (2)$$

20 Then

$$\frac{dQ}{dt} = \left[\frac{\partial f}{\partial F_1} \frac{dF_1}{dt} + \frac{\partial f}{\partial F_2} \frac{dF_2}{dt} + \frac{\partial f}{\partial T_I} \frac{dT_I}{dt} + \frac{\partial f}{\partial T_R} \frac{dT_R}{dt} + \frac{\partial f}{\partial T_O} \frac{dT_O}{dt} + \frac{\partial f}{\partial y_1} \frac{dy_1}{dt} + \frac{\partial f}{\partial y_2} \frac{dy_2}{dt} \right] \quad (3)$$

Since dQ/dt is the rate of change of Q with change in t , hence at fixed time intervals

25 $\Delta Q = Q(n) - Q(n-1) = \frac{dQ}{dt} \text{ for } t = T, \text{ where } T = \text{control action interval.}$

Thus:

(4)

$$\Delta Q = \left[\frac{\partial f}{\partial F_1} \Delta F_1 + \frac{\partial f}{\partial F_2} \Delta F_2 + \frac{\partial f}{\partial T_I} \Delta T_I + \frac{\partial f}{\partial T_R} \Delta T_R + \frac{\partial f}{\partial T_0} T_0 + \frac{\partial f}{\partial y_1} \Delta y_1 + \frac{\partial f}{\partial y_2} \Delta y_2 \right]$$

where:

$$5 \quad \frac{\partial f}{\partial F_1} = - \left[c_{P_i} (T_R - T_I) \right] / \lambda \quad (5)$$

$$\frac{\partial f}{\partial F_2} = y_1 \Delta H_1 + y_2 \Delta H_2 + c_{P_1} (T_R - T_0) / \lambda \quad (6)$$

$$\frac{\partial f}{\partial T_I} = F_1 c_{P_i} / \lambda \quad (7)$$

$$\frac{\partial f}{\partial T_R} = (F_2 c_{P_1} - F_1 c_i) / \lambda \quad (8)$$

$$\frac{\partial f}{\partial T_0} = -F_2 c_{P_1} / \lambda \quad (9)$$

$$10 \quad \frac{\partial f}{\partial y_1} = F_2 \Delta H_1 / \lambda \quad (10)$$

$$\frac{\partial f}{\partial y_2} = F_2 \Delta H_2 / \lambda \quad (11)$$

$$c_{P_i} = \sum_{k=1}^4 c_{P_k} x_k \quad (12)$$

$$c_{P_1} = \sum_{m=1}^4 c_{P_m} y_m \quad (13)$$

and, temperature dependence of specific heat and heat of reactions are given by;

15

$$\Delta H = \Delta H_0 + \int_0^T c_p dT; \text{ and}$$

$$c_p = a + bT + cT^2$$

A control system embodying the invention can thus be obtained where all measured signals are interfaced to a control computer system by state of the art methods. The total control system is shown in Figure 1.

5 Major steps of the control system are given in Figure 2. The calculations for a block B-9 are those which have been developed above. Calculations in all other blocks are based upon commonly known practices.

10 The outputs are: coolant flow control valve setting at 112; minimum and maximum temperature signals to start-up and shut-down control systems (not shown here), and to display units at 114; and high low temperature alarms at 116. Other measured signals can be checked for high and/or low limits and alarmed as per need and operating
15 practices of an individual reactor system.

 While the implementation shown herein is through a control computer system, the invention can also easily be implemented by conventional electronic instrumentation and control systems.

20 Referring once more to the drawings, a plurality of temperature transmitters 62 provide temperature signals to computer 20 for the reactor feed line, the reactor 10 at various longitudinal locations thereof and the reactor effluent line. Each of the analog signals is converted in
25 a corresponding analog to digital converter 63 into a corresponding digital signal which is readable by computer 20. In addition to the temperature transmitters, the system is provided with a plurality of flow transmitters 44 and a plurality of concentration transmitters 50. Flow transmitters
30 44 provide analog signals which similarly are converted to corresponding digital signals corresponding to the reactor feed and effluent flow rate as well as the coolant flow rate.

Concentration transmitters 50 provide signals corresponding to the concentration of various components in the feed and effluent line.

Each of the transmitters is further identified and correlated with inputs to the computer shown in Fig. 2. Temperature transmitters TT11 to TT1M provide their signals to a signal processing block B-2. The signals are then processed in a block B-4 to determine the maximum and minimum temperatures along the longitudinal length of the reactor. The signal is then supplied over line 14 to an output of computer 20 and also to the block B-9 for achieving calculations as pointed out above. Limit checking is accomplished in a block B-5 for sounding a low or high temperature alarm over line 116.

The transmitters and blocks identified above are all individually available in the art and will not be described in greater detail.

According to the present system, thus, not only is the coolant flow rate Q as set forth in equation (1) obtained, but also the change in coolant flow rate ΔQ . Equation (4) is utilized with each differential factor calculated as shown in equations (5) to equation (11). Simple comparators which relate an initial value with a later value can be utilized in computer 20 to obtain the values of flow rate change, ΔF_1 and ΔF_2 , as well as the temperature and concentration change.

CLAIMS

1. A system for controlling the temperature of a reactor (10) for containing a reaction from at least one reactant to at least another product, the reactor (10) having a feed line for the reactant and an effluent line for the product, the arrangement being characterised by:
 - 5 a feed flow transmitter (44) connected to the feed line for measuring the flow (F_1) of reactant to the reactor (10);
 - an effluent flow transmitter (44) connected to the effluent line for measuring the flow (F_2) of product from the reactor (10);
 - a feed temperature transmitter (62) connected to the feed line for
 - 10 sensing the reactant temperature (T_1);
 - an effluent temperature transmitter (62) connected to the effluent line for measuring the product temperature (T_O);
 - at least one reactor temperature transmitter (62) connected to the reactor for measuring a temperature (T_R) of the reactor;
 - 15 a concentration transmitter (50) connected to the effluent line for measuring the concentration of the at least one product in the effluent line;
 - a coolant flow line to the reactor (10) for supplying coolant to the reactor at a coolant flow rate;
 - 20 coolant flow control means (100) in the coolant line; and
 - circuit means that
 - (i) is connected to all of said transmitters (50, 62) and to the coolant flow control means (100) for controlling the flow of coolant to the reactor (10) according to a coolant flow signal,
 - 25 (ii) is connected to receive quantities proportional to the heat of reaction for at least one reaction in the reactor (10), specific heats of the reactant and product, and the heat of vaporization of the coolant,
 - (iii) is operable to obtain values for changes per unit time in feed flow rate (ΔF_1), effluent flow rate (ΔF_2), feed temperature (ΔT_1), reactor
 - 30 temperature (ΔT_R), effluent temperature (ΔT_O), and concentration of at least one product (Δy), and

(iv) includes circuit components for multiplying each change per unit time by a characteristic factor.

2. A system according to claim 1, wherein the circuit means includes circuit elements for generating factors as follows:

$$\frac{\partial f}{\partial F_1} = - \left[C_{P_i} (T_R - T_I) \right] / \lambda$$

$$\frac{\partial f}{\partial F_2} = y_1 \Delta H_1 + y_2 \Delta H_2 + C_{P_1} (T_R - T_0) / \lambda$$

$$\frac{\partial f}{\partial T_I} = F_1 C_{P_i} / \lambda$$

$$\frac{\partial f}{\partial T_R} = (F_2 C_{P_1} - F_1 C_i) / \lambda$$

$$\frac{\partial f}{\partial T_0} = -F_2 C_{P_1} / \lambda$$

$$\frac{\partial f}{\partial y_1} = F_2 \Delta H_1 / \lambda$$

$$\frac{\partial f}{\partial y_2} = F_2 \Delta H_2 / \lambda$$

the coolant flow signal being changed by a quantity ΔQ in the circuit means which is calculated as follows:

$$\Delta Q = \left[\frac{\partial f}{\partial F_1} \Delta F_1 + \frac{\partial f}{\partial F_2} \Delta F_2 + \frac{\partial f}{\partial T_I} \Delta T_I + \frac{\partial f}{\partial T_R} \Delta T_R + \frac{\partial f}{\partial T_0} \Delta T_0 + \frac{\partial f}{\partial y_1} \Delta y_1 + \frac{\partial f}{\partial y_2} \Delta y_2 \right]$$

wherein:

λ = coolant heat of vaporization;

y_1 = a first product concentration;

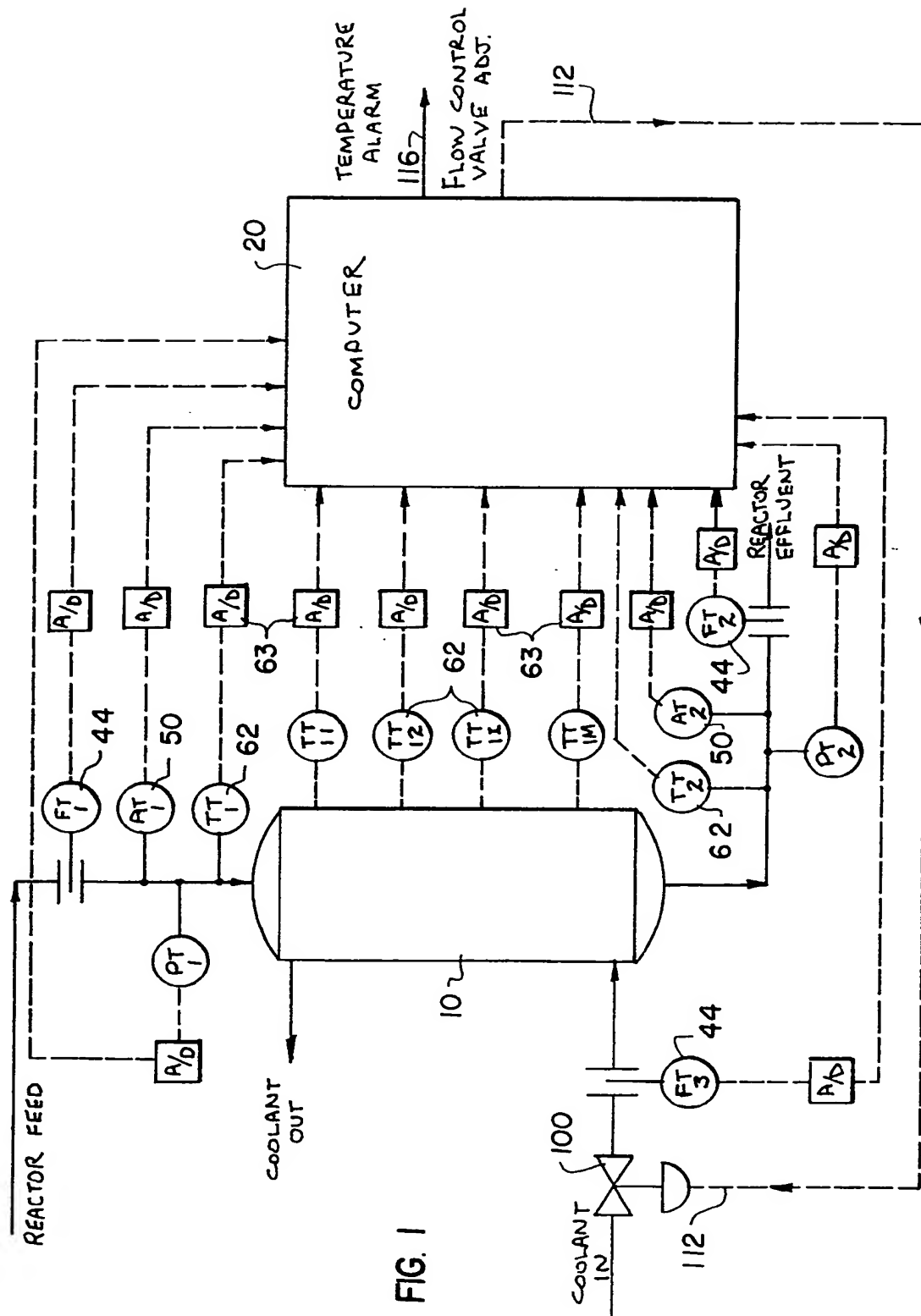
ΔH_1 = heat of reaction of reactant to first product;

5 y_2 = a second product concentration;

ΔH_2 = heat of reaction of reactant to second product; and

C_{p1} = specific heat of effluent.

3. A system according to claim 2, for a reactor (10) in which ethylene plus oxygen is supplied in use to the reactor as reactant and ethylene oxide plus carbon dioxide and water are generated as products, y_1 being the concentration of ethylene oxide, y_2 being the concentration of carbon dioxide, ΔH_1 being the heat of reaction of ethylene plus oxygen to ethylene oxide, and ΔH_2 being the heat of reaction of ethylene plus oxygen to carbon dioxide.
- 10
- 15 4. A system according to claim 1, claim 2 or claim 3, comprising a plurality of temperature transmitters (62) distributed along the length of the reactor (10), and a minimizing/maximizing circuit connected to the temperature sensors for obtaining a minimum and a maximum temperature among the temperature sensors of the reactor.



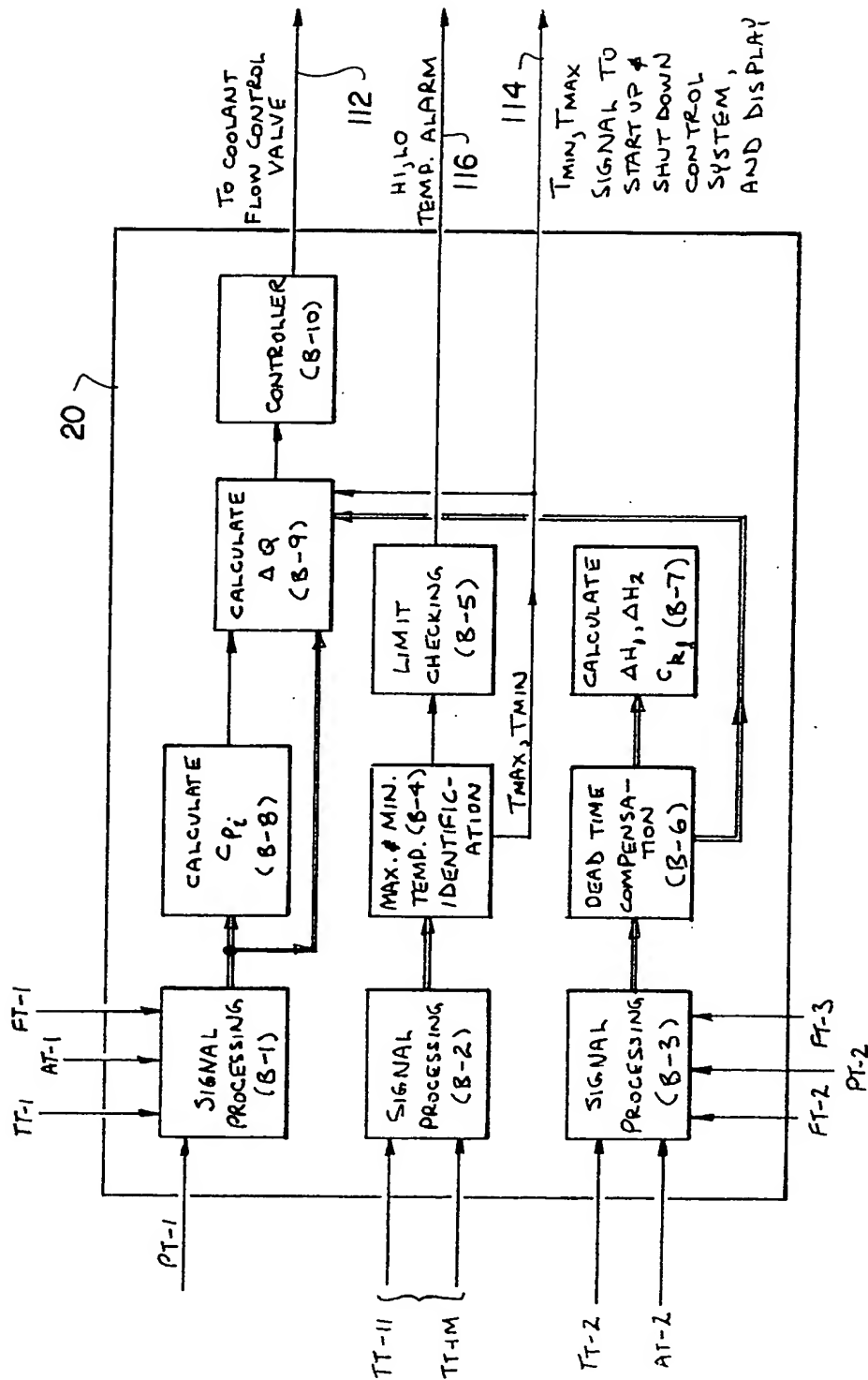


FIG. 2

This Page Blank (uspto)

(12)

EUROPEAN PATENT APPLICATION

(21) Application number: 83302539.8

(51) Int. Cl.: **B 01 J 8/06**
C 07 D 301/08

(22) Date of filing: 05.05.83

(30) Priority: 07.05.82 US 375795

(43) Date of publication of application:
16.11.83 Bulletin 83/46

(88) Date of deferred publication of search report: 10.07.85

(84) Designated Contracting States:
DE FR GB IT

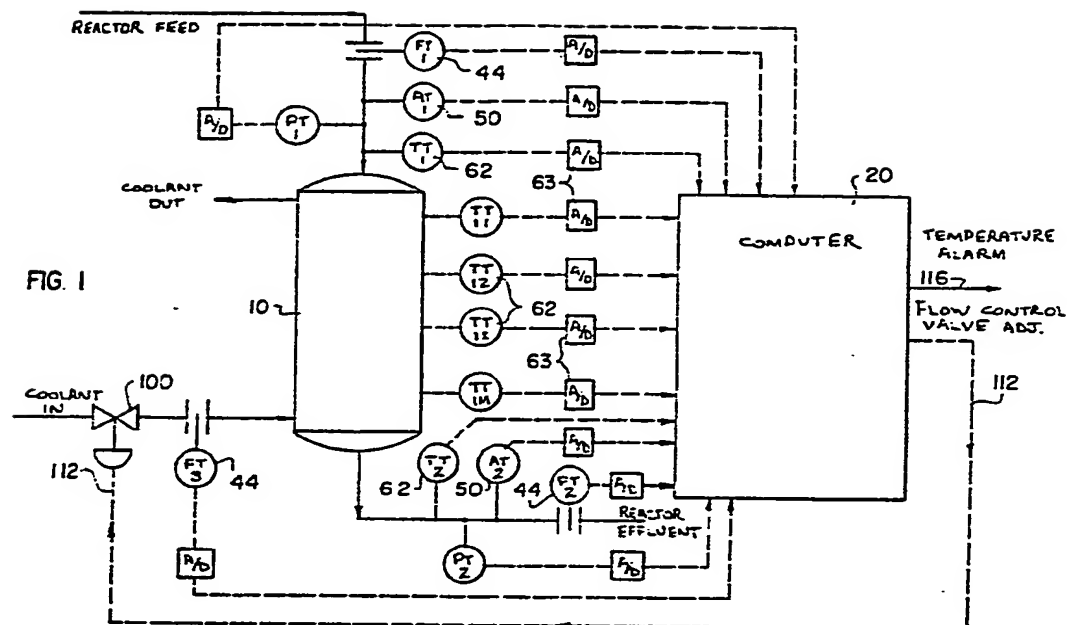
(71) Applicant: **THE BABCOCK & WILCOX COMPANY**
1010 Common Street P.O. Box 60035
New Orleans Louisiana 70160(US)

(72) Inventor: **Agarwal, Suresh C.**
26011 Lakeshore Boulevard
Euclid Ohio 44132(US)

(74) Representative: **Cotter, Ivan John et al,**
D. YOUNG & CO. 10 Staple Inn
London WC1V 7RD(GB)

(54) **Reactor temperature control systems.**

(57) A temperature control system controls the temperature of an olefin oxidation reactor (10) by measuring various parameters in the reactor and in flow lines to and from the reactor, and using these parameters with known quantities for specific heat of the feed and effluent and heats of vaporization and reaction, to calculate a coolant flow rate set point. Parameters relating to the heat balance with regard to a desired olefin oxide product and undesired carbon dioxide products are utilized for correct control. Equipment is also provided for ascertaining the state of a catalyst used in the reaction. Circuit components are provided for obtaining changes in flow rates, temperatures and concentrations to calculate a change in coolant flow rate as well as the steady state effluent flow rate quantity to improve the system to start up, shut down and transient conditions.





European Patent
Office

EUROPEAN SEARCH REPORT

0094208
Application number

EP 83 30 2539

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl. 3)
A	CHEMICAL ABSTRACTS, vol. 92, no. 16, 21st April 1980, page 148, no. 131449a, Columbus, Ohio, US; & SU - A - 706 101 (LENINGRAD TECHNOLOGICAL INSTITUTE) (30-12-1979)	1	B 01 J 8/06 C 07 D 301/08
A	FR-A-2 356 197 (I.C.I.) *Claims 1-4, 8, 10-14*	1	
A	US-A-4 236 218 (W.N.KILLEBREW, Jr. et al.) *Abstract; figures 1, 2*	1	
A	US-A-3 271 472 (F.T.OGLE et al.) *Claim 3; figure*	1	
A	US-A-4 249 908 (G.L.FUNK) *Abstract; figures 1, 2*	1	
A	FR-A-2 318 674 (DART INDUSTRIES) *Claims 1, 2; figures 1, 4, 5*	1, 4	
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 13-03-1985	Examiner SIEM T.D.
CATEGORY OF CITED DOCUMENTS			
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document	

EPO Form 1503 03 82

THIS PAGE BLANK (USPTO)